



Coupled Mechanical-Reactive Transport Modeling of Damage of Cementitious Materials due to Accelerated Decalcification

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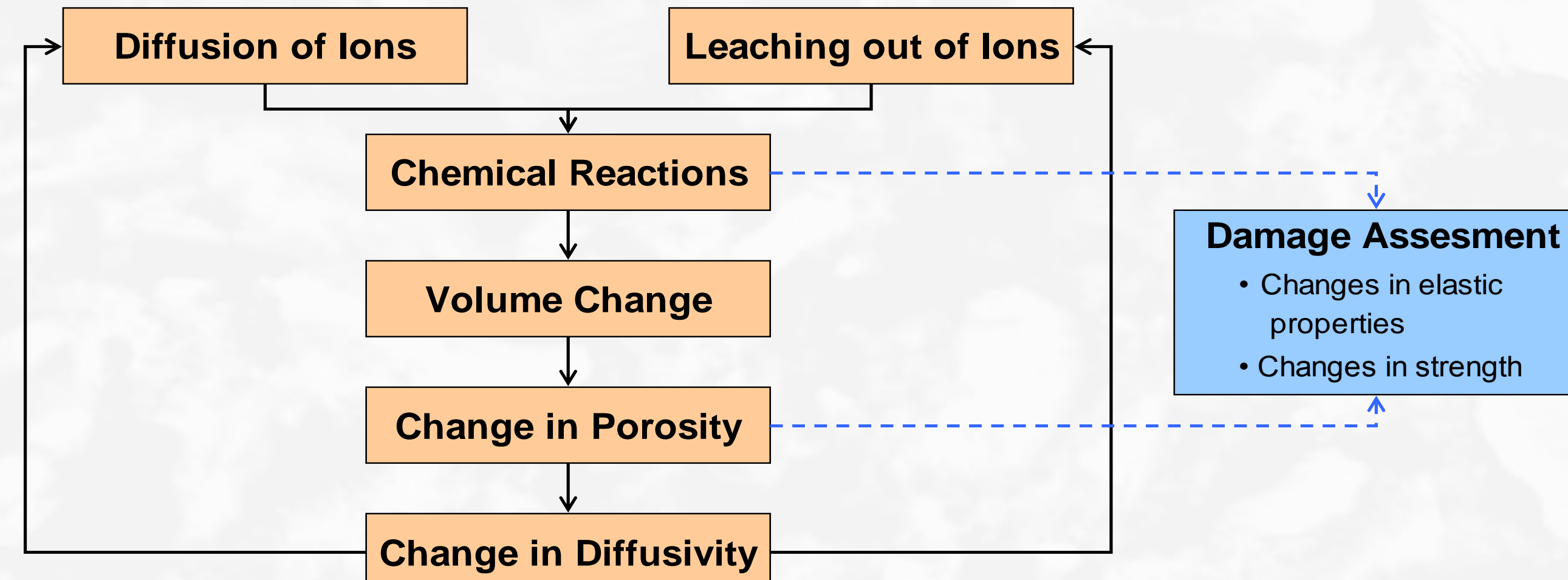
I. Motivation

- Cement-based materials are subjected to a variety of environmental weathering forces that cause internal chemical changes which result in an increase in localized stresses and a decrease in the mechanical properties
- Thorough understanding of the impact of degradation on the service life of cementitious materials is needed in order to design more sustainable structures
- Lab degradation tests can be lengthy in time and require a large number of samples, therefore a model which can couple the chemo-mechanical processes of cementitious materials exposed to aggressive environments is needed

II. Objectives

- Develop a numerical simulation framework capable of coupling the chemical reactions due to exposure to aggressive environments with changes in the mechanical properties of the material
- Simulate the influence of accelerated decalcification on the physical and mechanical properties of a Portland cement paste

III. Modeling Framework



Mineral Set Initialization

- Use of geochemical speciation code and pH dependent test data to generate a typical mineral set for a Type I/II Portland cement

III. Modeling Framework, cont'd.

Diffusion of Ions

- Use of diffusion equation in saturated porous media under isothermal conditions

$$\frac{\partial(\varphi c)}{\partial t} = \text{div} \left[\frac{D^0 \varphi}{\tau(\text{grad}(c))} \right]$$

c = concentration of an ion
 D^0 = free solution diffusivity of an ion
 φ = porosity
 τ = tortuosity

Chemical Reactions

- Use of ORCHESTRA, a geochemical speciation and transport code, to calculate the equilibrium phases of the major material constituents

Changes in Porosity

- Use of changes in volume relationship to account for solid phase dissolution/precipitation in the matrix pores

$$\varphi = \varphi_0 - \frac{\Delta V_s}{V}$$

φ = current porosity
 φ_0 = initial porosity
 V = volume of the representative volume element
 ΔV_s = change in the solid volume

Changes in Diffusivity

- Use of Sampson and Marchand's porosity/volume relationship¹ to account for changes in porosity

$$D = D^0 \frac{e^{\frac{4.3\varphi}{V_p}}}{e^{\frac{4.3\varphi_0}{V_p}}}$$

V_p = volume of the paste

Damage Assessment

- Determine homogenized stiffness matrix (elastic moduli)

- Four homogenization stages

Stage I: C-S-H

Stage II: cement paste (homogenized C-S-H from Stage I, Portlandite, ettringite, gypsum, etc.)

Stage III: cement mortar (sand particles/capillary pores in a homogenized cement paste from Stage II)

Stage IV: concrete (coarse aggregate in a homogenized cement mortar from stage III)

- Two homogenization schemes

- Self-consistent approach (Stages I and II)
- Mori-Tanaka approach (Stages III and IV)

- Determine homogenized Young's modulus

$$\bar{E} = \frac{9\bar{k}\bar{\mu}}{3\bar{k} + \bar{\mu}}$$

\bar{k} = bulk modulus
 $\bar{\mu}$ = shear modulus

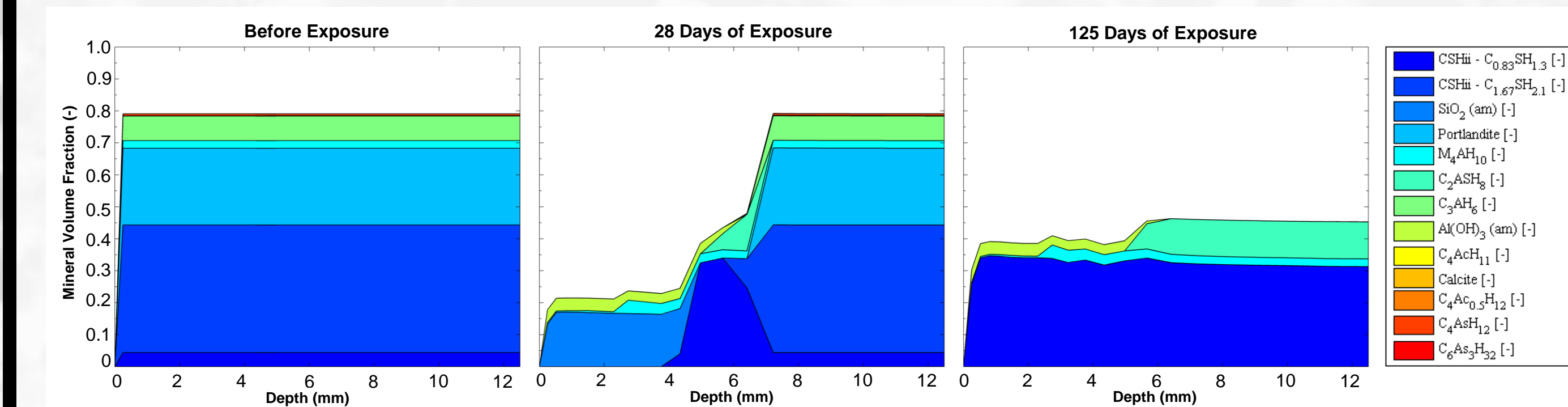
¹ Samson, E. and Marchand, J. Modeling the transport of ions in unsaturated cement-based materials. Computers & Structures, 2007, 85(23-24), p. 1740-1756.

IV. Results

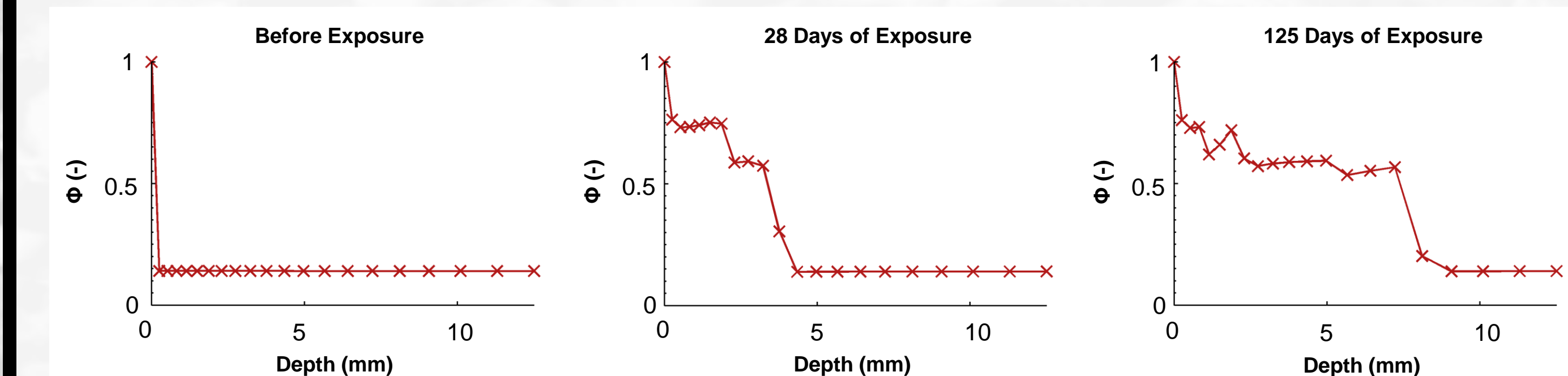
Case Study

- Portland cement paste (Type I/II)
- Prismatic samples exposed to 6M ammonium nitrate solution for 7, 14, 28, and 125 days
- For simulations, prismatic sample idealized as an 80 element one-dimensional structure

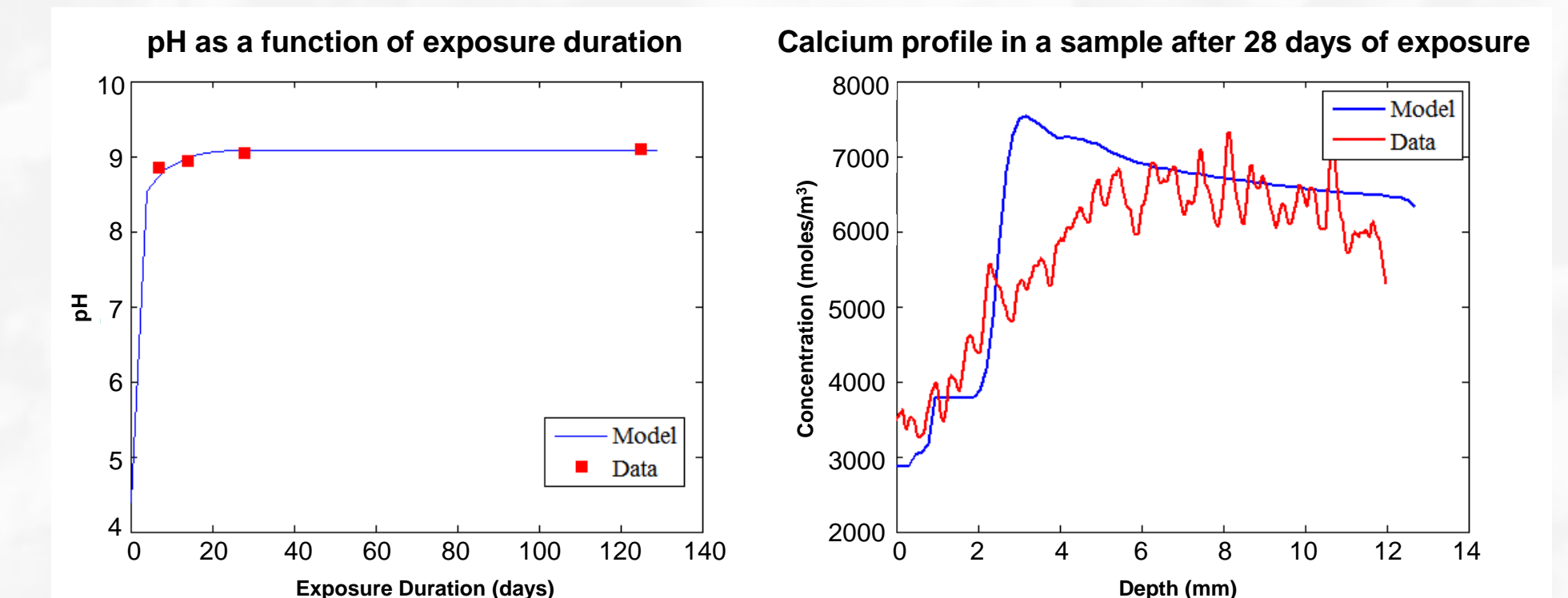
Mineral Volume Fractions



Porosity



Comparison with Experimental Data



V. Future Modifications

Modifications to Accommodate Inclusions

- Dual regime model
 - Diffusion through immobile zone
 - Advection through mobile zone
 - Exchange between zones through diffusion

